## Fast antibacterial kinetics of metal oxides coated polymers

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Compact uniform and adhesive  $TiO_2$  deposited on thin polymer PE-films have been prepared by DC-magnetron sputtering (DCMS).  $TiO_2$  sputtered on polyethylene (PE) showed effective and fast bacterial reduction kinetics and methylene blue (MB) self-cleaning properties under low intensity solar light [1-2].



During the bacterial inactivation, the shift in the vibrational peaks of the infrared  $-CH_2$  bands was attributed to the increase in the  $-CH_2$  bond stretching taking place preceding bond lysis and complete bacterial inactivation. The bacterial inactivation time was concomitant with the time required for the hydrophobic to hydrophilic transformation on PE-TiO<sub>2</sub> surface under band-gap irradiation [3]. The production of malondialdehyde (MDA) was observed during *E. coli* loss of viability.

The first evidence for Cu–Ag (50%/50%) nanoparticulate hybrid coatings is presented leading to a complete and almost instantaneous bacterial inactivation in the dark ( $\leq$ 5 min). Dark bacterial inactivation times on Cu–Ag (50%/50%) were observed to coincide with the times required by actinic light irradiation on 3D polyurethane catheters. This provides the evidence that the bimetal Cu-Ag driven inactivation predominates over a CuO/Cu<sub>2</sub>O and Ag<sub>2</sub>O oxides inducing a semiconductor driven behavior. Cu- or Ag-coated polyurethane catheters led to bacterial inactivation needing about ~30 min.

These  $PE-TiO_2$  and other metals/metal oxides sputtered surfaces present a potential practical application for the disinfection since they preclude the formation of biofilms on PE, polyurethane and medical textiles [4-5].

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